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Characterization of a freezing/melting transition in a vibrated and sheared granular medium

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Abstract. We describe experiments on monodisperse spherical particles in an annular cell geometry, vibrated from below and sheared from above. This system shows a freezing/melting transition such that under sufficient vibration a crystallized state is observed, which can be melted by sufficient shear. We characterize the hysteretic transition between these two states, and observe features reminiscent of both a jamming transition and critical phenomena.

Keywords: phase diagrams (experiment), granular matter, slow relaxation and glassy dynamics, fluctuations (experiment)

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1. Introduction

Granular materials exhibit phases analogous to conventional solids, liquids, and gases, in spite of being athermal and dissipative [1]. Due to the dissipation, energy must be supplied in order to sustain a dynamical state. Shearing and vibration are two common means to inject energy into granular systems. Shearing a granular material can compact and crystallize it [2], but also melt it [3]; tapping will compact it [4]; in thin vibrated layers there can be coexistence of crystallized and disordered states [5]; and highly vibrated granular systems become gas-like. From a large phase space of variables we vary only two, the shear rate and vibration amplitude, and study the interaction of the two energy injection mechanisms.

Without vibration, sheared granular materials undergo a phase transition from solidlike to fluid-like behaviour: the particles must become unjammed (which typically involves dilation) before they can move. We seek to understand what effects vibrations have on such transitions, and on the characteristics of the states on either side of the transition. This is particularly interesting given that granular systems are athermal, and one might naively expect that vibrations would play a temperature-like role.

We perform experiments in a classic geometry, annular shear flow [6]-[9], with monodisperse particles, shown schematically in figure 1. Shear and vibration provide competing effects, with the system evolving to a crystallized state when the kinetic energy provided by the vibration is greater than that provided by the shear. The transition is hysteretic, and fluctuations in the packing fraction and the breadth of the force distribution both become large as the crystallized state is approached, in similarity to phase transitions in other systems.

The physical parameters that characterize the system include the amplitude A and frequency f of vibration, the height h and mean radius r of the annular container, the diameter d and the density ρ of the particles, the rotation rate Ω of the upper shearing surface, and the mean pressure P on the layer (here characterized at the base of the layer).

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Figure 1. Schematic cross-section of experiment (not to scale).

Г	$\frac{A\omega^2}{g}$	vibrational acceleration/gravitational acceleration
Η	$\frac{h}{d}$	cell height/particle size
Ι	$\frac{\dot{\gamma}d}{c}$	particle scale velocity/acoustic velocity
J	$\frac{\Omega r}{A\omega}$	apparatus scale velocity/vibration velocity
K	$\frac{A\omega}{c}$	vibration velocity/acoustic velocity
L	$\frac{A}{d}$	vibration length scale/particle length scale
M	$\frac{\dot{\gamma}}{\omega}$	vibration timescale/shear timescale
N	$\frac{P}{\rho g d}$	applied pressure/hydrostatic pressure
R	$\frac{r}{d}$	cell radius/particle diameter
$\tilde{\Omega}$	$\frac{\Omega r}{\sqrt{gd}}$	shear velocity/particle velocity

Table 1.	. Dimensionless	ratios,	with ω	$\equiv 2\pi f,$	$\dot{\gamma} \equiv 9$	$\Omega r/h$,	and $c \equiv $	'P/	ρ.
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From these physical parameters, it is possible to define other dimensioned parameters, such as the shear rate, $\dot{\gamma} = \Omega r/h$, as well as a number of dimensionless parameters which we list in table 1.

 Γ , *I* and *J* are three key parameters from the list in table 1. Of the ten listed, there are only seven independent parameters: for example, $\Gamma = K^2 N/L$ and the four velocity ratios (H, I, J, K) only represent three parameters. In the experiments described here, we have fixed *f*, *P*, and *N* and therefore only explored a small region of the available phase space.



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(b)

Figure 2. Sample images, viewed from outer wall. (a) Crystallized state: $\Gamma = 2.0$ and $\Omega = 0.078$. (See also movie cryst.mpg.) (b) Disordered state: $\Gamma = 2.0$ and $\Omega = 0.47$. Linear (L) and hexagonal (H) clusters marked by black boxes. (See also movie disord.mpg.) (c) Phase diagram for crystallized and disordered states as a function of Ω and Γ . Dashed line is J = 1. Adapted from [3].

2. Experiment

(a)

The experimental apparatus consists of an annular region containing nearly monodisperse polypropylene spheres of diameter d = 2.29-2.39 mm and density $\rho = 0.90$ g cm⁻³, as shown in figure 1, with the pressure P and volume V (height h) set from below by a spring within an electromagnetic shaker. The particles are sheared from above and vibrated from below, while the sidewalls are stationary. The boundary conditions for the shakercontrolled bottom plate allow for dilation of the system under shear, with fast vibrations (A = 0 to 0.2d) superimposed upon slow dilation (maximum ~0.5d). The bottom plate surface is smooth aluminium, and the top plate has a disordered layer of glued particles. A more detailed description of the apparatus is given in [3]. To characterize the states, we obtain high-speed video images of particles at the outer Plexiglas wall, laser position measurements of the bottom plate (cell volume), and force time series from a capacitive sensor flush with the bottom plate. For the experiments described in this paper, we fix the frequency of vibration (f = 60 Hz) and number of particles ($N \approx 71200$), and vary the amplitude of vibration A and shear rate Ω . We vary the nondimensionalized peak acceleration $\Gamma \equiv A(2\pi f)^2/g$ from 0 to 6, and the nondimensional shear rate $\Omega \equiv \Omega r/\sqrt{gd}$ from 0.058 to 9.3.

3. Description of states

In the regime $0 < \tilde{\Omega} < 10$ and $0 < \Gamma < 6$ we observe two distinct granular states of matter: crystallized and disordered. Sample images and movies of these two states are shown in figures 2(a) and (b), as viewed from the outer wall. For $\Omega \leq 1$ and $\Gamma < 4$, we observe that the phase boundary between the two states roughly corresponds to a curve where the characteristic velocities of the two motions are equal. This corresponds to the





Figure 3. Force probability distribution functions for three values of $\tilde{\Omega}$ at $\Gamma = 2.0$: (a) on linear scales, dimensioned and (b) on log-linear scales, normalized by mean force. See figure 6 for the dependence of $\langle F \rangle$ on $\tilde{\Omega}$.

dimensionless number J of table 1:

$$J \equiv \frac{\Omega r}{2\pi f A} \tag{1}$$

and the curve J = 1 is shown by the dashed line in figure 2(c). Below, we characterize these two states, with further details to be found in [3].

Crystallized state. In the solid-like state (see figure 2(a)), the balls crystallize into a hexagonally close-packed (HCP) configuration (i.e. a 3D crystalline structure) here visible only at the outer wall although the order persists throughout. We observe a packing fraction $\phi = 0.69$, which is lower than $\phi_{\text{HCP}} = 0.74$ due to defects, the curved geometry, and error in the measurement of ball and cell dimensions. The contact between the upper layer of the granular material and the shearing wheel is intermittent, with stick-slip motion of the top ~2 layers in the manner of [10]. The distribution of forces measured at the bottom of the layer is bimodal (see figure 3); while there is some asymmetry in the two peaks, this indicates that the material is largely responding as a solid body moving up and down with the sinusoidal vibrations of the bottom plate.

Disordered state. In the disordered state, some order remains in the form of hexagonally packed clusters and linear chains of particles at the outer wall, as marked in figure 2(b). For states with Ω well above the transition, linear chains dominate over hexagonal clusters, with both existing intermittently throughout the disordered regime. These chains may correspond to the planar ordering reported recently by Tsai *et al* [2, 11]. The velocity profile extends deeper into the layer (in the vertical direction) than in the crystallized state. Force distributions measured at the bottom plate show the exponential-like tails characteristic of many granular experiments in disordered, unvibrated granular materials (see figure 3). They also fall to zero at low force, as seen in earlier experiments by Miller *et al* [7].

For a geometrically similar system, but unvibrated, with rough lower surface, and exposed to a compressional force, shear *ordered* the system into horizontal planes of hexagonal packing, each slipping past the others [2, 11]. Such a state is different from the 3D crystallized state observed here, in which the layers in the bulk are stationary

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with respect to each other. An interesting question is how shearing creates order or disorder depending on the presence or absence of vibration. A useful way to distinguish the ordered and disordered regimes is via the ratio

$$I \equiv \frac{\dot{\gamma}d}{\sqrt{P/\rho}} \tag{2}$$

involving the shear timescale $\dot{\gamma} \equiv \Omega r/h$ to the acoustic timescale, $\sqrt{d^2 \rho/P}$, calculated from the pressure P and density ρ [12]. The experiments described in this paper have pressures of around P = 20 Pa and shear rates of $\dot{\gamma} = 0.3$ –40 Hz, leading to $I = 5 \times 10^{-3}$ to 0.75. In [2], the shear rates are slower, $\dot{\gamma} = 0.05$ –0.5 Hz, and pressures are higher, P = 2000 Pa, so that the system is clearly in the quasistatic regime with $I = 3 \times 10^{-5}$ to 3×10^{-4} .

While it is perhaps surprising that we find as simple a result as a phase transition at $J \approx 1$, the presence of these other important control parameters give hints into the breakdown of crystallization for large Γ . Figure 2(c) shows that for $\tilde{\Omega} \approx 0.7$, crystallization was not observed above $\Gamma = 4$, possibly indicating the re-emergence of disorder due to granular-gas-like behaviour. Further experiments varying (L, M, N) will be necessary to discover which of these determine the high- Γ boundary of the crystallized phase. In addition, the parameter H controls finite size effects.

4. Shear localization

Granular materials commonly exhibit shear banding, with an exponentially decaying velocity profile away from the shearing surface. As shown in figure 4(a), this shear band behaviour is seen in both the crystallized ($\tilde{\Omega} = 0.087$) and disordered states ($\tilde{\Omega} = 0.87$) for $\Gamma = 2.0$. To obtain the velocity profiles, we tracked individual particles visible at the outer wall using a high-speed video camera and determined trajectories for each. We then binned the resulting velocity components by depth to construct velocity profiles.

We characterize the azimuthal velocity, v(y), by fitting the profile to the form

$$v(y) = v_{\infty} + \alpha v_{\parallel} e^{-y/y_0} \tag{3}$$

where v_{∞} is the shear-induced sliding velocity at the bottom plate, v_{\parallel} is the known azimuthal velocity of the shear wheel at the outer wall, α is the efficiency with which that velocity is transmitted to the top layer of granular material, and y_0 is the decay length of the velocity with depth y.

In the crystallized state, the shear is localized almost entirely to the first layer of particles (small y_0), while in the disordered state the shear band extends several particles into the layer. The slip at the upper plate is lowest in the disordered state, where the uppermost particles are in constant contact with the shearing wheel. Note that the system is more dilated in the disordered state, and has a larger pressure [3]. While the disordered state has greater slip (v_{∞}) at the bottom plate, the scaled slip values $(v_{\infty}/v_{\parallel})$ are in fact lower than in the crystallized state, visible in figure 4(a). For disordered states with $\tilde{\Omega} \gtrsim 0.4$, the shear bands appear to have reached a steady state since they are all parameterized by the same values.





Figure 4. Characterization of velocity profiles at $\Gamma = 2.0$ and various Ω . (a) Azimuthal velocity measured at outer wall as a function of depth measured. Lines are fits to equation (3). (b) Solid body rotation v_{∞} as a function of $\tilde{\Omega}$, (c) decay length y_0 as a function of $\tilde{\Omega}$ and (d) efficiency coefficient α as a function of $\tilde{\Omega}$.

5. Transition

We examined the transition from the disordered to the crystallized state by first preparing a disordered state at high $\tilde{\Omega}$, then adjusting $\tilde{\Omega}$ to the value of interest. We then performed two runs, one at constant $\tilde{\Omega} = 0.27$ (starting from $\Gamma = 0$) and the other at constant $\Gamma = 2$ (starting from $\tilde{\Omega} = 8.4$). The mean volume measured for each step of these two runs is shown in figure 5. Γ was held approximately constant by fixing the amplitude of the signal driving the shaker. As can be seen in figure 2(c), the resulting vibration amplitude varied less than 5% based on the state of the material.

For steps of decreasing Ω (figure 5(a)) the system compacts logarithmically until reaching Ω_c , after which the system undergoes a first-order phase transition to the crystallized state. While the discontinuity corresponds to only a 1% change in the volume of the cell, this represents 20% of the overall change observed. Below Ω_c only a small amount of additional compaction occurs, to a state with a volume V_{\min} , for which the packing fraction is $\phi = 0.69$. When Ω is increased, the transition back to the disordered state is hysteretic, occurring for $\Omega_h \approx 2\Omega_c$.

For steps of increasing Γ (figure 5(b)) the system also compacts. However, runs approaching the transition are difficult to repeat quantitatively, since there is a great deal





Figure 5. Volume V of cell, scaled by minimum observed volume V_{\min} as a function of (a) $\tilde{\Omega}$ (at $\Gamma = 2.0$) and (b) Γ (at $\tilde{\Omega} = 0.27$). Triangles point in direction of steps.

of intermittency in the cell volume (see figures 5(b), 8(a) and [3] for details). For $\Gamma > \Gamma_c$ the system is in the crystallized state. The transition also appears to be first order, but in this case the hysteresis is so extreme that the material was not observed to re-expand when we decreased Γ .

We wish to understand why a crystallized state can disorder by increasing Ω , but not by decreasing Γ . In the case of increasing Ω , the stick-slip behaviour in the top layers of the crystallized state is affected by the speed of the upper shearing wheel. As Ω increases, more horizontal momentum is transferred to the upper layer of balls, which results in longer regions of flowing particles. Eventually, the whole layer can be seen to mobilize and the disordering begins to take place throughout the cell. In contrast, for increases in A (and hence Γ) no such increased momentum transfer takes place, and the results are similar to the irreversibility observed for compaction by tapping [4]. This transition shows some similarity to the 'freezing-by-heating' transition seen in [13], in which individual particles with tunable noise are seen to crystallize as their noise level is increased. Such a system also shows hysteresis in returning to the disordered, mobilized state.

For the run at $\Gamma = 2.0$, seen in figure 5(a) with steps downward in Ω , we observe signatures of the phase transition from disorder to crystallization via both the volume fluctuations and the force distribution, as shown in figures 6 and 7. As $\Omega \to \Omega_c$ from above, both the volume fluctuations (measured from the variance of V(t)) and the breadth of the force distribution (measured by the kurtosis, or fourth scaled moment, of F(t) on the force sensor) become large. In addition, the first-order nature of the transition is visible in other characteristics of the force distribution, such as the mean, standard deviation, and skewness (see also figure 3).

In granular systems, Edwards and co-workers [14] have introduced a temperature-like measure, the compactivity, defined as $X = (\partial V/\partial S)_N$ by analogy with thermodynamics. The central idea is that lower packing fractions correspond to a greater freedom for particle rearrangement, and hence a higher compactivity. In the statistical mechanics of ordinary second-order phase transitions, susceptibilities such as $(\partial^2 A/\partial T^2)_V$ (for free energy A) are singular at the critical point. For example, the specific heat at constant volume is





Figure 6. Characteristics of force probability distributions as a function of $\tilde{\Omega}$ at $\Gamma = 2.0$. Triangles point in direction of steps in $\tilde{\Omega}$; solid points are crystallized phase; the dashed line is a guide to the eye.

 $C_V = (\partial E/\partial T)_V = -T(\partial^2 A/\partial T^2)_V$. When described in terms of fluctuation-dissipation relations, $k_{\rm B}T^2C_V = \langle (E - E_0)^2 \rangle$, where E is the energy of the system and E_0 its mean value. One expects energy fluctuations, and hence C_V , to be singular at the critical temperature T_c . By contrast, at a first-order transition, discontinuities occur in densities, but one does not expect divergent fluctuations. Since V has taken the place of E in the Edwards formalism, the hallmark of a critical transition is increased fluctuations in the volume of the system as we approach X_c , the critical compactivity. In our experiments, volume (and hence X) is set by Ω , and the inset to figure 7 shows apparently singular behaviour for the volume fluctuations as a function of the volume. It is interesting that, in these experiments, we see a discontinuity in the density but also an indication of a singularity in the volume (density) fluctuations. The magnitude of the fluctuations observed in the disordered state is similar to those in observed in [15], where the standard deviation of the packing fraction is approximately 10^{-4} .

6. Intermittency

The apparently singular volume fluctuations near J = 1 come from the fact that the system exhibits intermittency in its state. The system is in fact spatially inhomogeneous, with instances of small V being crystallized in the majority of the cell and instances of large V being majority disordered. By examining the properties of the system in this intermittent regime, we are able to compare a broad range of states for nearly the same parameter values. The only varying parameters are the volume and pressure of the system,

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Figure 7. Volume fluctuations as a function of $\hat{\Omega}$, scaled by average volume V_0 . Inset: volume fluctuations as a function of V_0/V_{\min} at each $\tilde{\Omega}$. Triangles point in direction of steps in $\tilde{\Omega}$; solid points are crystallized phase; the dashed line is a guide to the eye.



Figure 8. Characterization of velocity profiles in intermittent regime at $\hat{\Omega} = 0.27$ and $\Gamma = 1.0$. (a) Azimuthal velocity measured at outer wall as a function of depth measured for compact (\bigcirc , t = 3.8 h) and dilated (\square , t = 4.7 h) states. Lines are fits to equation (3). (b) Solid body rotation v_{∞} as a function of cell height, (c) decay length y_0 as a function of cell height and (d) efficiency coefficient α as a function of cell height. Solid symbols correspond to data from (a).

which are related to each other by a proportionality constant due to the spring constant of the shaker [3].

The inset to figure 8(a) shows a time series of the cell height for such a run, with $\Gamma = 1.0$ and $\tilde{\Omega} = 0.27$. The system started from a dilated state and progressed to a majority crystallized state before re-dilating and re-compacting to an even more crystallized state over the course of approximately 10 h. We again obtain velocity profiles at the outer wall, in this case while simultaneously monitoring the position of the bottom plate.

Figure 8(a) shows two velocity profiles from compact and dilated states. Because the system is spatially inhomogeneous, the particles in view of the camera may in fact be either disordered or crystallized at any given time during these measurements, regardless of the height of the cell. Importantly, the fit parameters in figures 8(b)–(d) show the same trends as those in figure 4 when the rotation rate $\tilde{\Omega}$ is taken as a proxy for cell height. In both cases, we observe a continuum of states as the system moves between crystallization (compaction) and disorder (dilation). Again, it is interesting to note that the volume and pressure fluctuations are associated with the formation and melting of ordered clusters over time. Such behaviour is characteristic of near-critical behaviour. By contrast, at a thermodynamic first-order transition, we would not expect to see persistent dominant fluctuations.

7. Discussion

The two characterizations of a transition in the system we discuss above provide contrasting, but complementary, information about the nature of the crystallizing phase transition in sheared and vibrated granular materials. The canonical hallmark of a transition to a jammed/glassy state is the continuous growth of the viscosity. Glass transitions do not in general contain first-order-like signatures, such as discontinuities in the volume or specific heat [16]. For sheared colloids, there are large stress fluctuations near a jamming transition [17], and in simulations of Lennard-Jones particles, force PDFs are observed to broaden [18, 19]. Similar behaviour is observed in this system as well, but with a density discontinuity. By contrast, jammed/glassy states are all *disordered*, while the granular system described in this paper makes a transition to a *crystallized* state. In both the glassy and crystallized cases, however, the final states are unable to rearrange.

We observe similarities to critical phenomena in the increased volume fluctuations near the transition, a hallmark at odds with a glass transition. These fluctuations are similar to the density fluctuations observed at the liquid–gas critical point, which occur at diverging length scales. Therefore, further investigations into the nature of this transition should examine what length scales and order parameters are present, including a determination of the sizes of clusters and the spatial correlations between forces. Finally, we have introduced a number of dimensionless control parameters whose effects remain to be investigated in future studies.

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